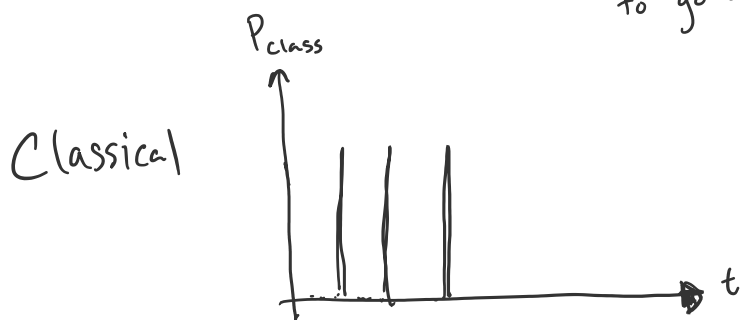
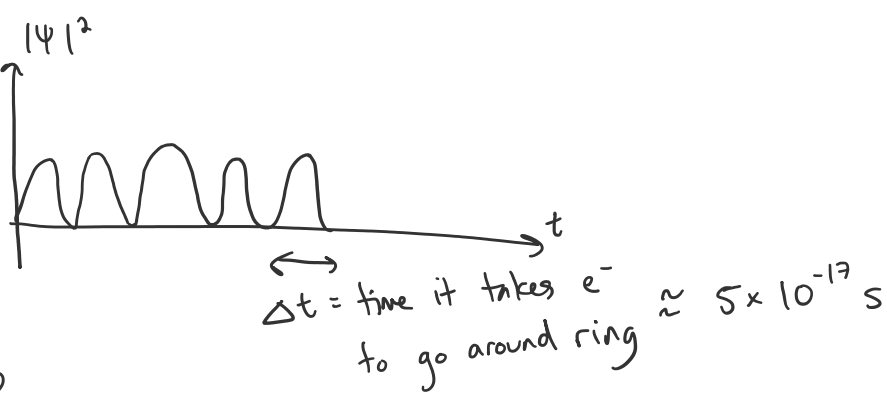
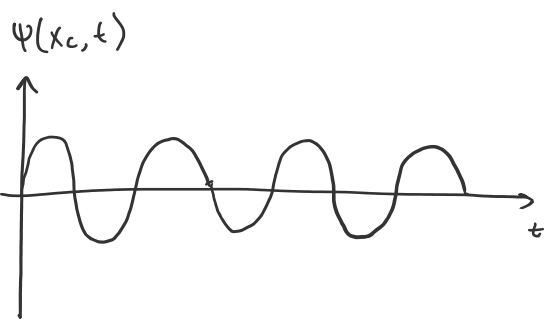
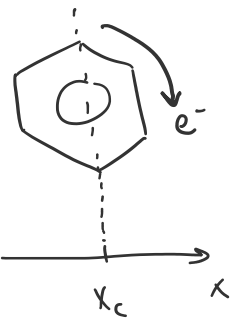


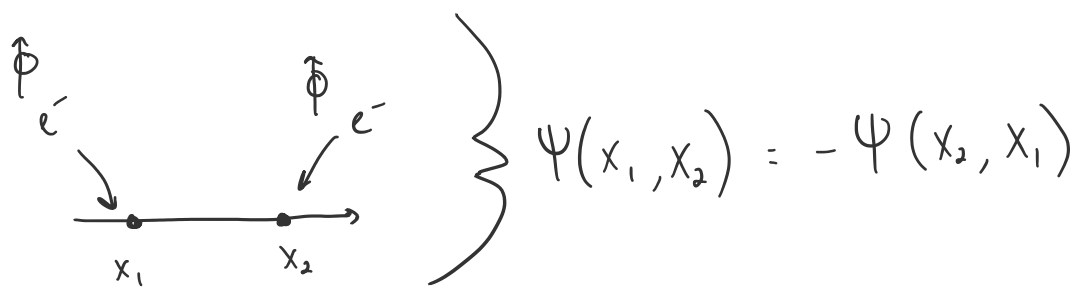
# Lecture 6

Friday, September 1, 2023 9:59 AM

ex #1 Time Dependence of wavefunction of an  $e^-$  going around a benzene ring



ex 2: Derive Pauli Exclusion Principle from P4



As the  $e^-$  approach each other meaning  $x_2 \rightarrow x_1$ ,  
 in the limit where  $x_2 \rightarrow x_1$ ,  
 $\Rightarrow \Psi(x_1, x_1) = -\Psi(x_1, x_1) \Rightarrow \Psi(x_1, x_1) = 0$

There is no probability for  $e^-$  occupying the same spot,  
 & probability drops as they approach  
 PEP explains why objects are solid

ex 3: Schrödinger equation

$$P1 \Rightarrow E = i\hbar \frac{\partial}{\partial t}$$

$$p = -i\hbar \frac{\partial}{\partial x} \Rightarrow p^2 = -\hbar^2 \frac{\partial^2}{\partial x^2}$$

$$P2 \Rightarrow \mathcal{H}\Psi = i\hbar \frac{\partial}{\partial t} \Psi$$

$$\left( \frac{p^2}{2m} + V(x) \right) \Psi = i\hbar \frac{\partial}{\partial t} \Psi$$

$$\left( \frac{-\hbar^2}{2m} \frac{\partial^2}{\partial x^2} + V(x) \right) \Psi = E\Psi \Rightarrow \mathcal{H}\Psi = E\Psi$$

ex 4: The order of things matters in QM

$$CM = x \cdot p, p \cdot x \Rightarrow xp - px = 0$$

$$QM = (xp - px)\Psi = \left( x \frac{\hbar}{i} \frac{\partial}{\partial x} - \frac{\hbar}{i} \frac{\partial}{\partial x} x \right) \Psi$$