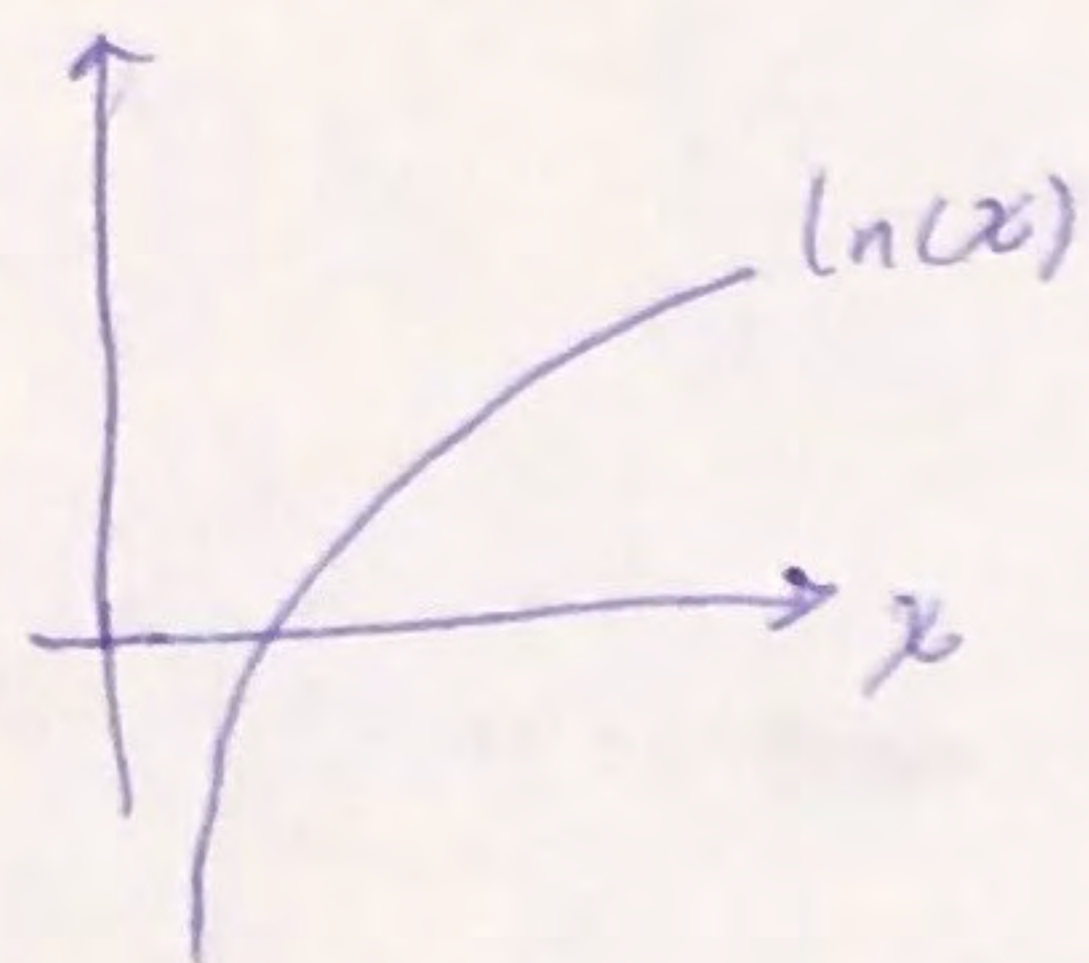


L4 - Review

math topics ...

⑤ $\ln(x)$



properties:

monotonic \Rightarrow unique
 compressed \Rightarrow great for
 small and large
 numbers

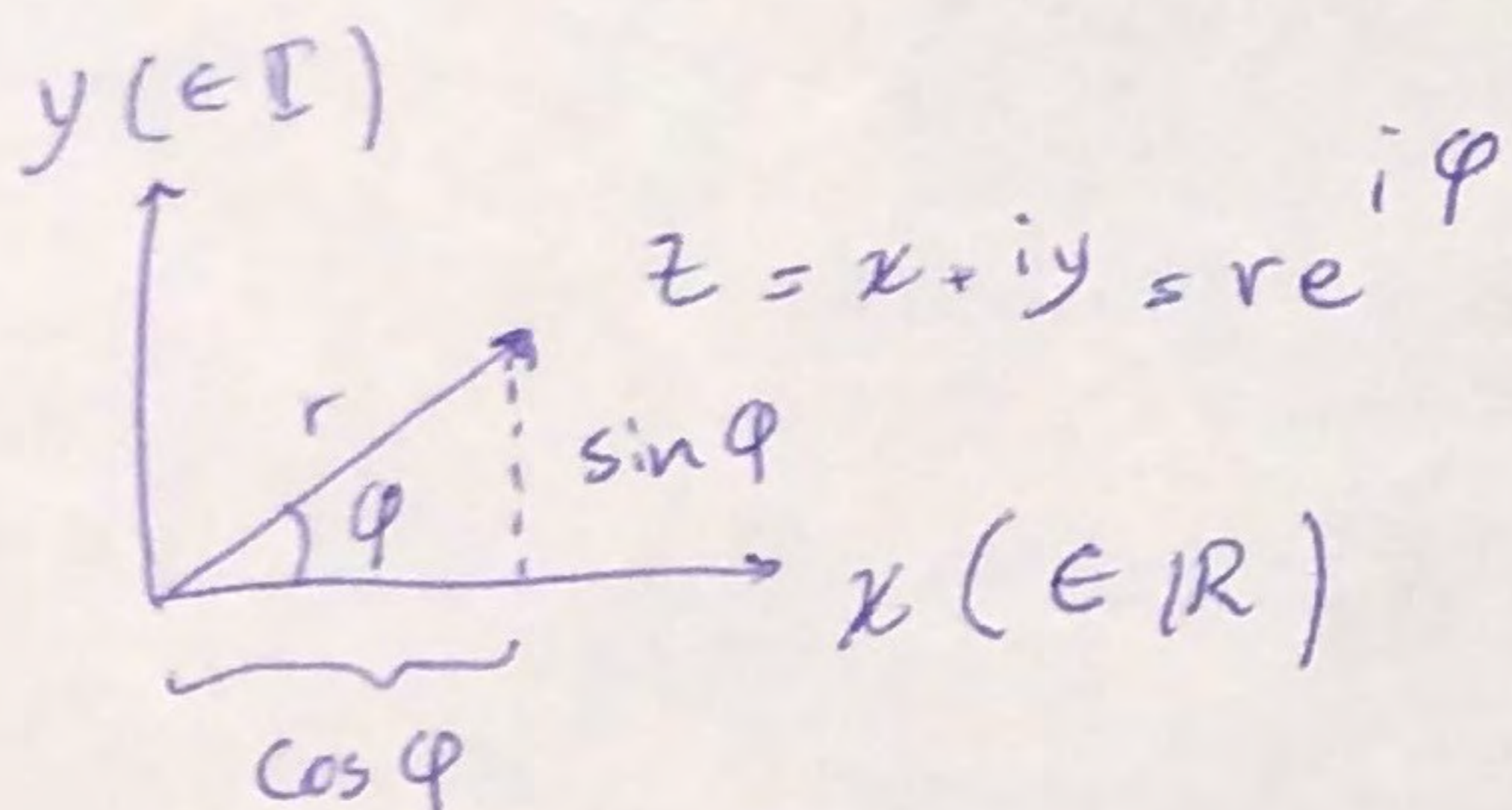
$$\ln(a \cdot b) = \ln(a) + \ln(b)$$

turns multiplicative \Leftrightarrow
 into additive, e.g.

$$w_1 \cdot w_2 \xrightarrow{\ln} S_1 + S_2$$

⑥ complex numbers

$$e^{i\varphi} = \cos \varphi + i \sin \varphi ; i \equiv \sqrt{-1}$$



ex: for i , $r = 1$; $\varphi = 90^\circ$ (or $\frac{\pi}{2}$)

$$(i = 1 \cdot e^{i \frac{\pi}{2}})$$

\downarrow \downarrow
 r φ

Today - Why go quantum?

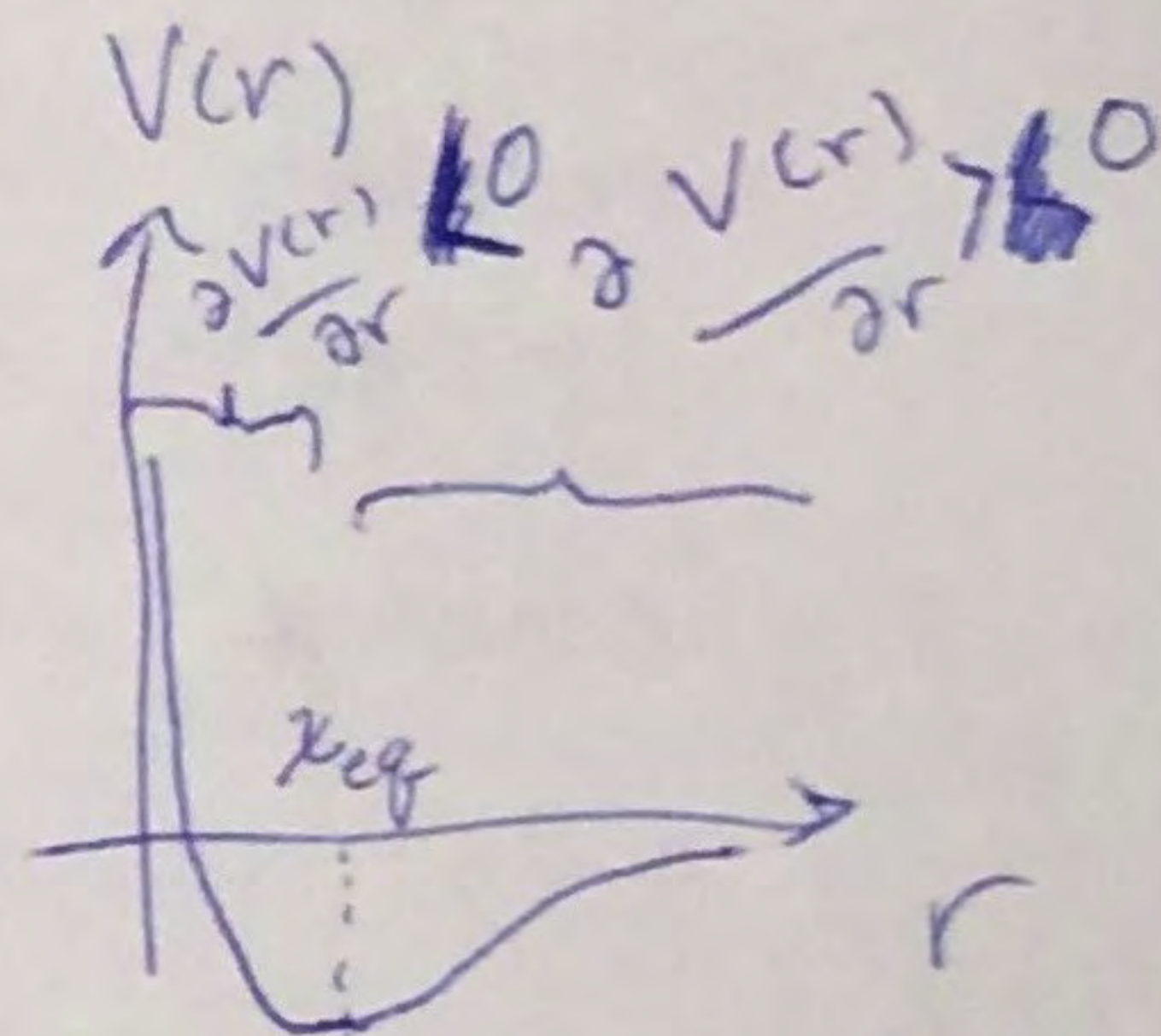
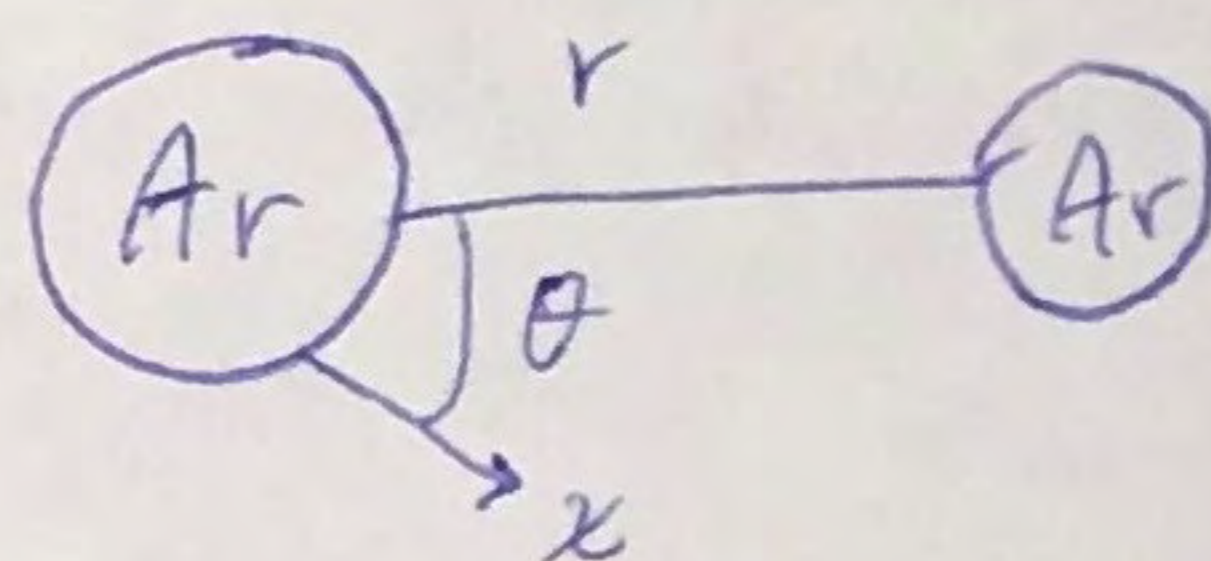
In classical mechanics, energy of particles

$$E = \underbrace{\sum \frac{1}{2m_i} P_i^2}_{\text{kinetic energy}} + \underbrace{V(x_i)}_{\text{potential energy}}$$

just an index, not $i = e^{i\pi/2}$

P_i : momentum

ex:



$$F = - \frac{\partial V}{\partial r}$$

force \downarrow

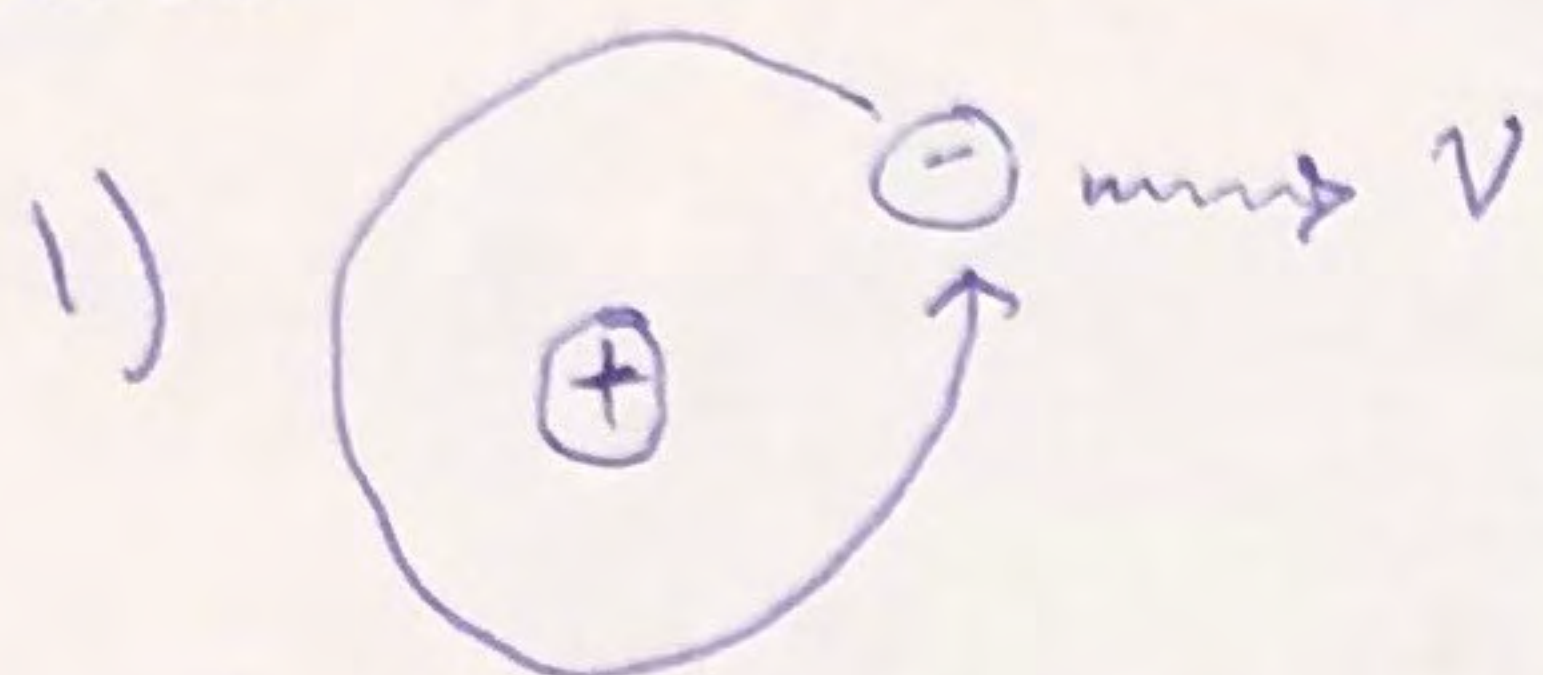
$$F = ma = m \frac{\partial^2 x}{\partial t^2}$$

$$\Rightarrow - \frac{\partial V}{\partial r} = m \frac{\partial^2 x}{\partial t^2}$$

\Downarrow
 if $r > r_{eq}$ then
 the two atoms
 will attract each
 other.

If $r < r_{eq}$ the
 Ar atoms will repel
 each other.

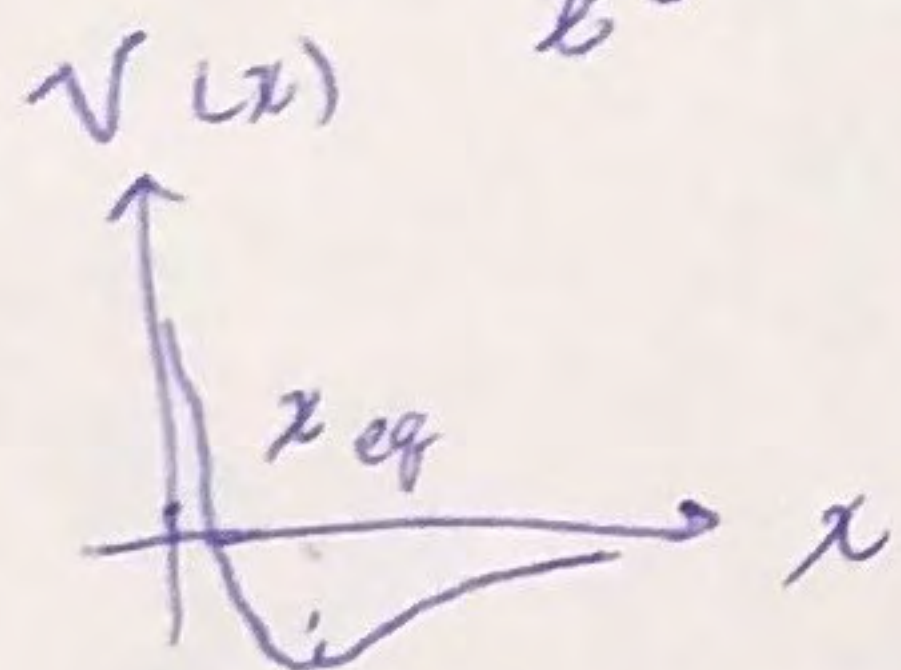
Problems with classical mechanics:



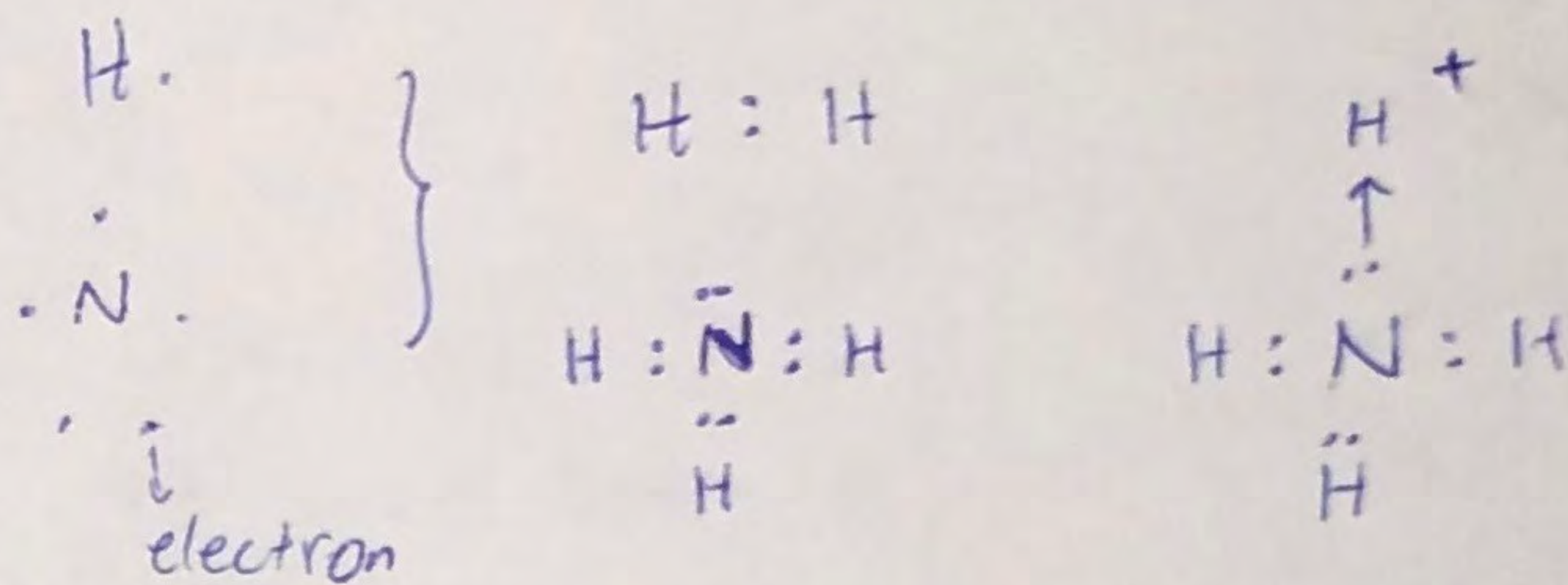
Classical mechanics: the negative charge is oscillating (the velocity is changing as it is orbiting) so according to electromagnetic laws it will emit light. Since light emission reduces the energy of the particle, it gets closer to the positive charge and it will eventually collapse \Rightarrow Paradox!

2) where do you get $V(x)$ from? (e.g. it is known empirically that

$$V \sim \frac{1}{x^6} \text{ for } x > x_{eq}$$



Early 1900s: GN Lewis:



Lewis's observation: in ~~most~~ (almost) all cases the electrons get paired so the total number of electrons is even

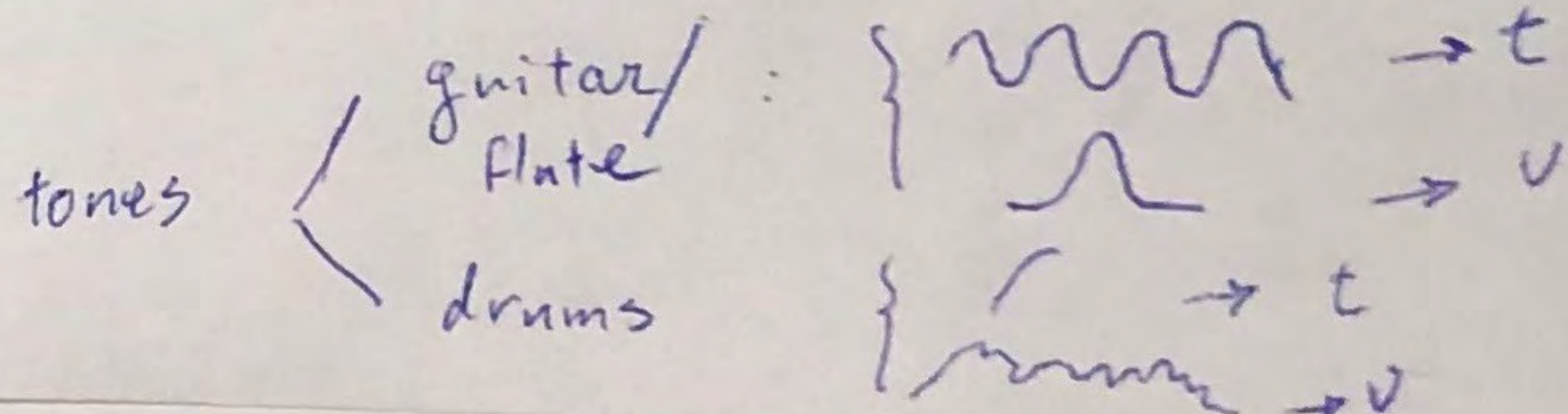
Heisenberg: 2 kinds of variables exist:

- 1) independent: can be measured with arbitrary precision simultaneously
- 2) conjugate.

ex: music

duration (t) frequency (v)

$$\Delta t \cdot \Delta v = \frac{1}{4\pi} ; \text{ it was known that there is an uncertainty in } v \text{ and } t$$



In classical mechanics: $\Delta x \cdot \Delta p = 0$
(x and p can be measured precisely
and simultaneously)

However, Heisenberg in 1925;

$$\Delta x \cdot \Delta p = \frac{h}{4\pi} ; h \approx 6.62 \times 10^{-34} \text{ J.s}$$

↓
Planck constant

h is so small \Rightarrow it was not discovered
until 1925.

Heisenberg's uncertainty has significance
only in very small systems, e.g. electrons