

Lecture 33

Last Time: Physico-chemical mechanics

Molecular motion equalizes $\mu(x_1)$ and $\mu(x_2)$ but how fast?

$P_1: \mu_{g_i} = \mu(\dots, x) \quad P_2 = J = -U_c \frac{du_{g_i}}{dx} = v_c$

Always true from math: $\frac{\partial J}{\partial x} = -\frac{\partial c}{\partial t}$

How to derive transport laws:

1) Find $\mu_g(x)$ for your system.

2) Plug into postulate 2: $J \neq 0$ out of equilibrium

$J = 0$ in equilibrium or steady state

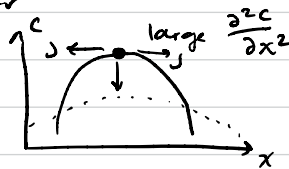
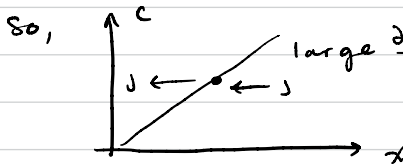
ex:



$\mu_g(x) = \mu^{(0)} + RT \ln c(x) \Rightarrow \frac{\partial c}{\partial t} = D \frac{\partial^2 c}{\partial x^2}$

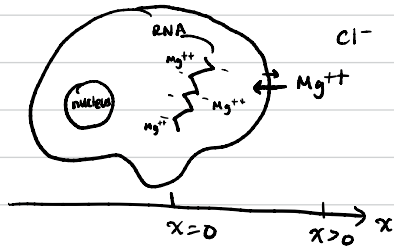
does not depend on position in beaker

Fick's law



Today: "Any law you want," part 1

ex: steady state of Mg^{++} in a cell



Mg^{++} binds to RNA and since $[RNA]_{cell} > [RNA]_{outside}$ the Mg^{++} conc will not be in equilibrium, but $J = 0$ when it reaches steady state inside and outside of the cell.

our system: $\mu_g(x) = \mu^{(0)}(x) + RT \ln c(x)$

$$\textcircled{2} \Rightarrow J \sim \frac{d\mu_g}{dx} = \frac{\partial \mu^{(0)}}{\partial x} + RT \frac{\partial \ln c(x)}{\partial x} = 0$$

$$\Rightarrow d\mu^{(0)} = -RT d(\ln c)$$

$$\Rightarrow \int_{\mu^{(0)}(x=0)}^{\mu^{(0)}(x)} d\mu^{(0)} = -RT \int_{c(x=0)}^{c(x)} d(\ln c)$$

$$\Delta \mu^{(0)} = \mu^{(0)}(x) - \mu^{(0)}(0) = -RT (\ln(c(x)) - \ln(c(0)))$$

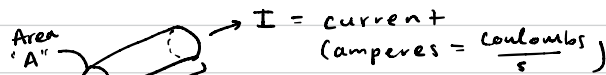
$$\Delta \mu^{(0)} = -RT \ln\left(\frac{c(x)}{c(0)}\right)$$

$$\Rightarrow e^{-\Delta \mu^{(0)}/RT} = c(x) / c(0)$$

$$\Rightarrow c(x) = c(0) e^{-\Delta \mu^{(0)}/RT}$$

due to RNA binding

ex: Ohm's law



$$I \left(\frac{\text{Coulombs}}{\text{s}} \right) = J \left(\frac{\text{moles}}{\text{m}^2 \text{s}} \right) \cdot A (\text{m}^2) \cdot z \left(\frac{\text{Coulombs}}{\text{mol}} \right)$$

molar charge ~ 96000 for electrons

$$= v_{\text{drift}} \cdot c \cdot A \cdot z$$

proved (for a protein on a gel): $v_{\text{drift}} = z E U = \frac{z E}{\gamma}$

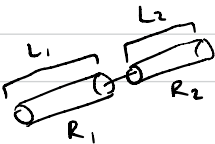
electric field $\frac{V_e}{L}$

$$\Rightarrow \text{combine } I = \frac{z U V_e}{L} \cdot A \cdot z$$

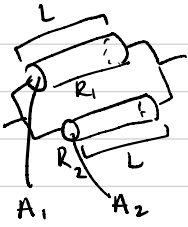
$$I = \frac{V_e}{R} \quad \text{w/ "resistance" } R = \frac{L RT}{z^2 D \cdot c \cdot A}$$

gas constant

"ohm's law"



$$\Rightarrow R = \frac{(L_1 + L_2) RT}{z^2 D c A} = \frac{L_1 RT}{z^2 D c A} + \frac{L_2 RT}{z^2 D c A} = R_1 + R_2$$



$$\frac{1}{R} = \frac{\epsilon^2 D C A}{L R T} = \frac{\epsilon^2 D C A_1}{L R T} + \frac{\epsilon^2 D C A_2}{L R T}$$
$$\frac{1}{R} = \frac{1}{R_1} + \frac{1}{R_2}$$