

Last Time: Transport

Isolated system: $F = ma$  Valid

Open system: $F = F_{\text{Applied}} + F_{\text{Friction}} = 0$ 

$$\Rightarrow - \frac{du}{dx} + F_{\text{Friction}} = 0$$

$\Rightarrow - \frac{du}{dx} = \gamma v$ if we neglect the ~~friction~~ viscosity force

ex: $V_{\text{drift}} = \frac{\Sigma Q}{\gamma}$ for protein on gel.

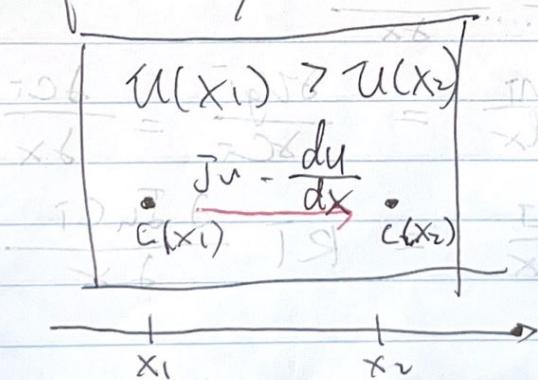
$$\text{Flux} = J \left(\frac{\text{mole}}{\text{m}^2 \cdot \text{s}} \right) = T \cdot c = - \frac{1}{\lambda} \frac{du}{dx} \cdot c$$

Also: P.531 always true

$$-\frac{dI}{dx} = \frac{dc}{dt}, \text{ reason: } \begin{array}{c} \text{C} \\ \xrightarrow{J} \\ \xleftarrow{J} \end{array} \quad \begin{array}{c} \vec{v} \cdot \vec{J} > 0 \\ \vec{v} \cdot \vec{J} > 0 \end{array}$$

two \vec{v} antiparallel also $\frac{dJ}{dx} > 0$.

Today: Postulate of physicochemical mechanics, or "How do molecules move in an open system?"



Flux moves particle to lower chemical potential.

For ideal solutions, $u = u^{(0)} + RT \ln C$

$C(x_2)$ increase $\Rightarrow C(x_1)$ decrease until $u(x_1) = u(x_2)$

P1: $u_i(T, P, n_i, \dots) \rightarrow u_{gi}(T, P, n_i, \dots, x)$

Assume: we only talk about volume δV large enough so $n \gg \sqrt{N}$.

P2: $J_{x\bar{x}} = -u_{gi} C_i \frac{\partial u_{gi}}{\partial x}$

κ "Flux" κ "molar force"

P1 $\Rightarrow \frac{\partial u_{gi}}{\partial x} = \frac{\partial u_{gi}}{\partial T} \cdot \frac{\partial T}{\partial x} + \frac{\partial u_{gi}}{\partial P} \cdot \frac{\partial P}{\partial x}$

\hookrightarrow molar free energy $+ \frac{\partial u_{gi}}{\partial n_i} \frac{\partial n_i}{\partial x}$

$G = \sum u_i n_i + \dots$

We can also simplify the 1st & last term in this eq.

Assume ideal system:

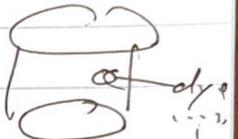
$u_{gi} = u_{gi}^{(0)} + RT \ln C$

$\Rightarrow \frac{\partial u_{gi}}{\partial x} = \frac{\partial u_{gi}^{(0)}}{\partial x}$

$\Rightarrow \frac{\partial u_{gi}}{\partial n_i} \frac{\partial n_i}{\partial x} = \frac{\partial u_{gi}^{(0)}}{\partial C_i} = \frac{\partial C_i}{\partial x}$

$= \frac{RT}{C_i} \frac{\partial C_i}{\partial x} = RT \frac{\partial \ln C_i}{\partial x}$

$$\Rightarrow \frac{\partial U_g}{\partial x} = \frac{\partial U_i^{(0)}}{\partial x} - S_T \frac{\partial T}{\partial x} + V_T \frac{\partial P}{\partial x} + \cancel{RT} \frac{\partial \ln C_i}{\partial x}$$



Example of deriving a "law".
 Let $U_g(x) = U_i^{(0)} + RT \ln C(x)$,
 one component, only concentration depends on x . $x = \frac{1}{C} = \frac{RT}{D}$

$$\Rightarrow \frac{\partial U_g}{\partial x} = RT \frac{\partial \ln C}{\partial x} = \frac{RT}{C} \frac{\partial C}{\partial x}$$

$$J = -U_C \frac{\partial U_g}{\partial x} = -URT \frac{\partial C}{\partial x} = -D \frac{\partial C}{\partial x}$$

$$\Rightarrow -\frac{\partial J}{\partial x} = D \frac{\partial^2 C}{\partial x^2}$$

$$\Rightarrow \text{since } -\frac{\partial T}{\partial x} = \frac{\partial C}{\partial t}$$

$$\frac{\partial C}{\partial t} = D \frac{\partial^2 C}{\partial x^2} \quad \text{Fick's law.}$$