

# L31: review

## Molecule random motion

a) Central Limit theorem

b)  $F_{\text{random}} \Rightarrow v$  is random  $\Rightarrow \Delta x$  is random

one-dimensional diffusion

$$\delta x = \sum_j \Delta x \Rightarrow \langle (\delta x(t))^2 \rangle = 2Dt$$

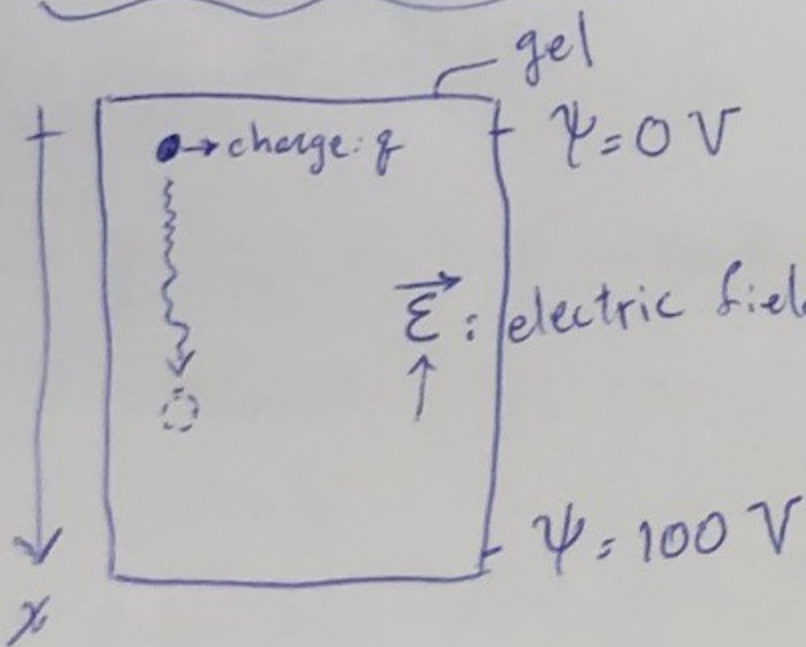
↓  
cumulative displacement

$$\frac{\langle \Delta x^2 \rangle}{\Delta t}$$

2-dimensional :  $\langle (\delta x(t))^2 \rangle = 4Dt$

3-dimensional :  $\langle (\delta x(t))^2 \rangle = 6Dt$

## Directed Motion



$$\psi = -x \cdot E$$

$$E = \nabla \psi + \dots + q \psi$$

$$M = \mu + \dots + q \psi$$

for the protein/DNA Particle moving on gel :  $F = F_{\text{random}} + F_{\text{friction}} + F_{\text{applied}} = ma$

Today

$$F \approx 0 \quad (\text{high friction})$$

①  $F_{\text{applied}} = -\frac{\partial E}{\partial x} = -\frac{\partial \mu}{\partial x}$

②  $F_{\text{friction}} \sim -v \Rightarrow F_{\text{friction}} = -\gamma v$

proportional to velocity because friction is caused by the number of collisions of particle with the gel matrix in the unit of time; faster movement causes more collisions.

① and ② :  $F = F_{\text{random}} - \gamma v - \frac{\partial \mu}{\partial x} = 0$

$$\Rightarrow \frac{\partial \mu}{\partial x} = F_{\text{random}} - \gamma v$$

Case 1:  $\mu = \text{constant} \Rightarrow \frac{\partial \mu}{\partial x} = 0 \Rightarrow v = \frac{F_{\text{random}}}{\gamma}$

$$\Rightarrow \frac{\partial x}{\partial t} = \frac{F_{\text{random}}}{\gamma} \Rightarrow \frac{\Delta x}{\Delta t} \approx \frac{F_{\text{random}}}{\Delta t}$$

$$\times \Delta x \Rightarrow \frac{\Delta x^2}{\Delta t} = \frac{F_{\text{random}} \cdot \Delta x}{\gamma} \rightarrow \Delta E_{\text{random}}$$

\* it can be shown that  $\Delta E_{\text{random}}$  is proportional to  $E_{\text{random}}$ . This is a result of linear response theory:

$$\Delta E_{\text{random}} \sim E_{\text{random}}$$

$$\left\langle \frac{\Delta x^2}{\Delta t} \right\rangle \sim \frac{E_{\text{random}}}{\gamma} \Rightarrow \left\langle \Delta x^2 \right\rangle \sim \frac{\langle E_{\text{random}} \rangle}{\gamma} \Delta t$$

$$\Rightarrow D \sim \frac{k_B T}{\gamma}$$

it can be shown that proportionality constant is  $\frac{k_B T}{\gamma}$

$$D = \frac{k_B T}{\gamma}; \quad \gamma = \frac{k_B T}{D} = \frac{1}{\mu}$$

$\downarrow$  friction coefficient       $\downarrow$  diffusion coefficient       $\downarrow$  mobility

Case 1

$$\Rightarrow \text{Diffusion law: } \langle \delta x(t)^2 \rangle = 2 \cdot \frac{k_B T}{\gamma} \cdot t$$

Case 2: Neglect random Force

$$\text{if } \mu = \mu_0 - \beta \cdot x \cdot E \Rightarrow \frac{\partial \mu}{\partial x} = -\beta \cdot E$$

$$\Rightarrow \frac{\partial \mu}{\partial x} = -\gamma v = -\beta E \Rightarrow v = \text{const} = v_{\text{drift}}$$

larger particles with equal charge drift slower because they have a higher friction coefficient.

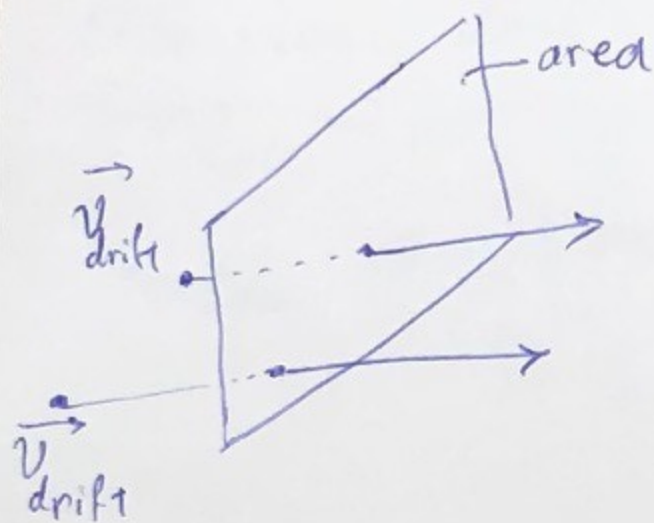
$$v_{\text{drift}} = \frac{\beta E}{\gamma}$$

Often (roughly spherical objects):

$$\gamma = \frac{1}{\mu} = 6\pi \cdot R \cdot \eta$$

radius of molecule  $\leftarrow$        $\hookrightarrow$  viscosity of gel (unit: Pa·s)

# Flux & continuity equation



Flux

$$J \left( \frac{\text{moles}}{m^2 \cdot s} \right) = v_{drift} \cdot C$$

Units:  $\frac{m/s}{m^2} \cdot \frac{\text{moles}}{m^3}$

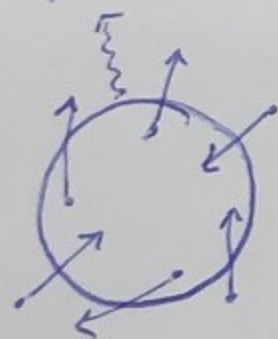
also:  $\frac{\partial M}{\partial x} = -\gamma v_{drift}$

$$v_{drift} = -\frac{1}{\gamma} \frac{\partial M}{\partial x} = -u \frac{\partial M}{\partial x};$$

$$J = -u C \frac{\partial M}{\partial x}$$

A: surface area

driving force, e.g.  $-qE$



$$\iiint dV \cdot \frac{\partial C}{\partial t} = \iint dA \cdot \vec{J}$$

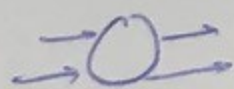
Gauss's theorem:  $\iint dA \cdot \vec{J} = \iiint dV \nabla \cdot \vec{J}$

divergence of flux



$$\nabla \cdot \vec{J} \neq 0$$

of flux



$$\nabla \cdot \vec{J} = 0$$

$$\Rightarrow \frac{\partial C}{\partial t} = \frac{\partial J}{\partial x} \quad (\text{continuity of flux})$$