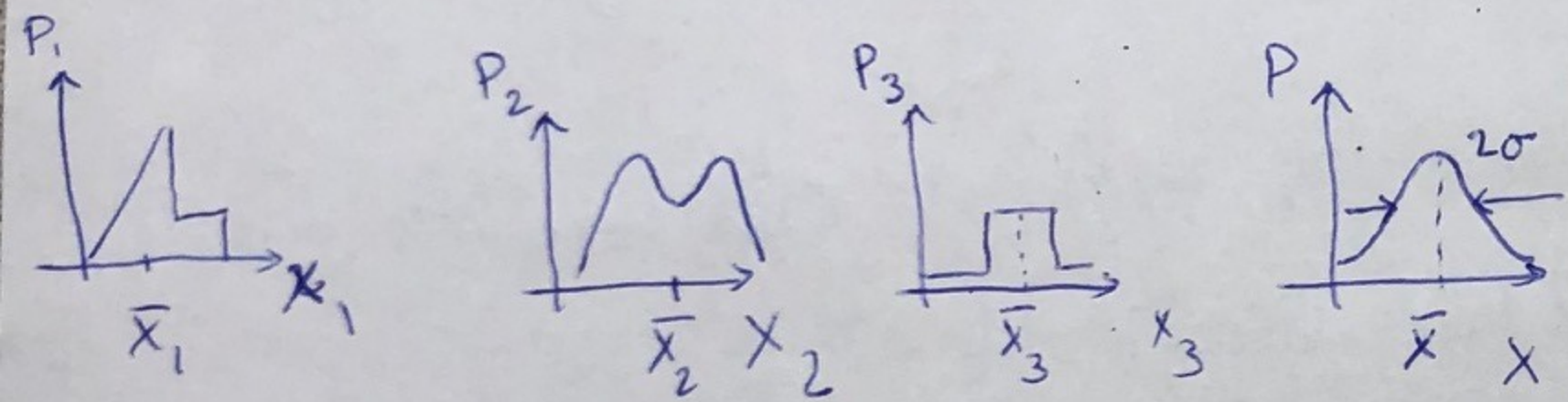


L30: review

Central limit theorem (C.L.T)

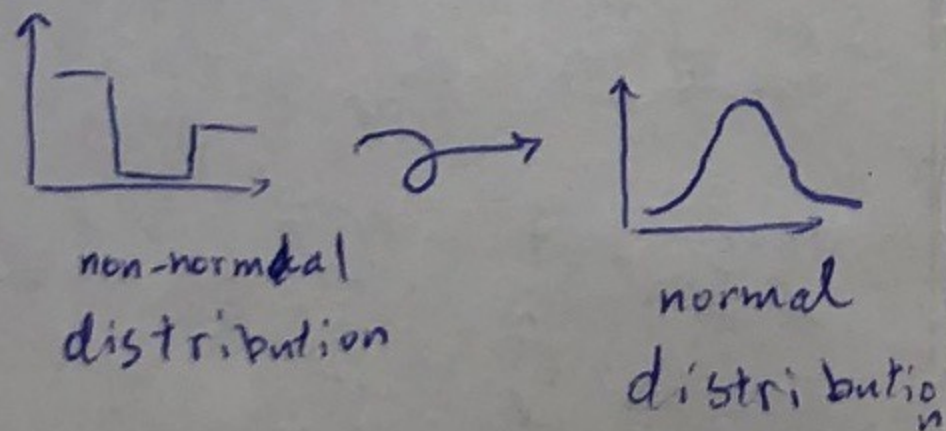
$$X_1 + X_2 + X_3 + \dots = X \quad \text{random variable}$$



X is gaussian distributed with $\bar{X} = \sum_j \bar{X}_j$ (average)
 $\sigma^2 = \sum_j \sigma_j^2$ (standard deviation)

* Prof. Gruebele showed a demo of non-normal distributions becoming normal if multiple numbers are picked from the non-normal distribution

and added together:
 (and this repeated thousands of time)



Today: How molecules move to get to equilibrium

Brownian motion

At short times, any motion is predictable by

Newtonian mechanics:

$$X(\Delta t) = X(0) + V(0) \cdot \Delta t$$

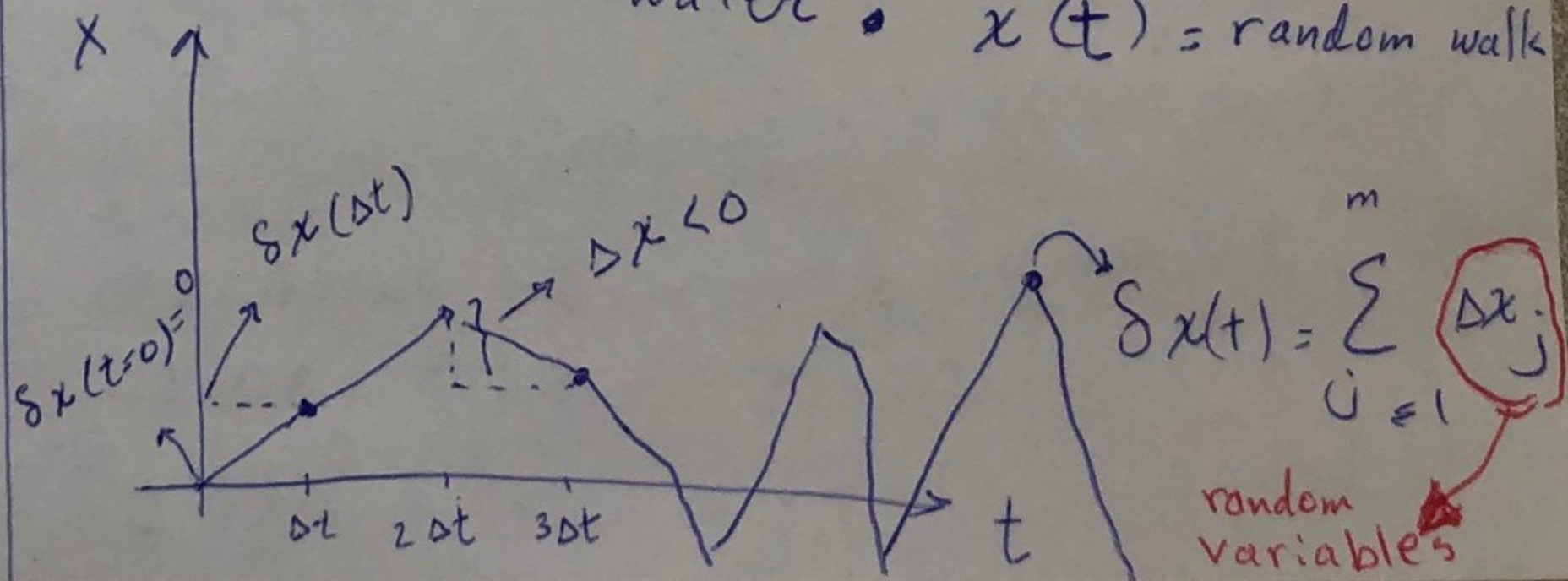
$$F = ma = m \frac{\Delta V}{\Delta t} \Rightarrow \Delta V = \frac{F}{m} \Delta t$$

$$\Rightarrow V(\Delta t) = \frac{F(0)}{m} \Delta t + V(0)$$

if F (from thermal motion) is random,

e.g. in the Brownian motion of a dust

particle in water: $x(t) = \text{random walk}$



No matter what the probability distributions of the Δx_j look like, for sure $\delta x(t)$ will be gaussian distributed after enough steps:

$$\delta x = \sum_{j=1}^m \Delta x_j, \quad m = \frac{t}{\Delta t}$$

$$\langle \delta x^2 \rangle = m \langle \Delta x^2 \rangle \quad \leftarrow \text{C.L.T}$$

$$\langle \delta x^2 \rangle = \frac{\langle \Delta x^2 \rangle}{\Delta t} \cdot t$$

Constant

$$\langle \delta x^2 \rangle = 2D \cdot t$$

⇒ root mean squared displacement $\sqrt{\langle \delta x^2 \rangle} \sim t^{1/2}$

D: diffusion coefficient

$$\begin{cases} 1\text{-dimensional diffusion: } \langle \delta r^2 \rangle = 2D \cdot t \\ 2\text{-} & \langle \delta r^2 \rangle = 4D \cdot t \\ 3\text{-} & \langle \delta r^2 \rangle = 6D \cdot t \end{cases}$$

Exercise: drug from patch has to diffuse through skin (e.g. nicotine patch)

$$\sqrt{\langle \delta x^2 \rangle} \sim 2 \text{ mm} \quad ; \quad D \sim 100 \mu\text{m}^2/\text{s}$$

how long does it ~~take~~ take for drug to diffuse through skin?

$$\langle 2000 \mu\text{m} \rangle^2 = 6 \cdot 100 \mu\text{m}^2 \text{s}^{-1} \cdot t$$

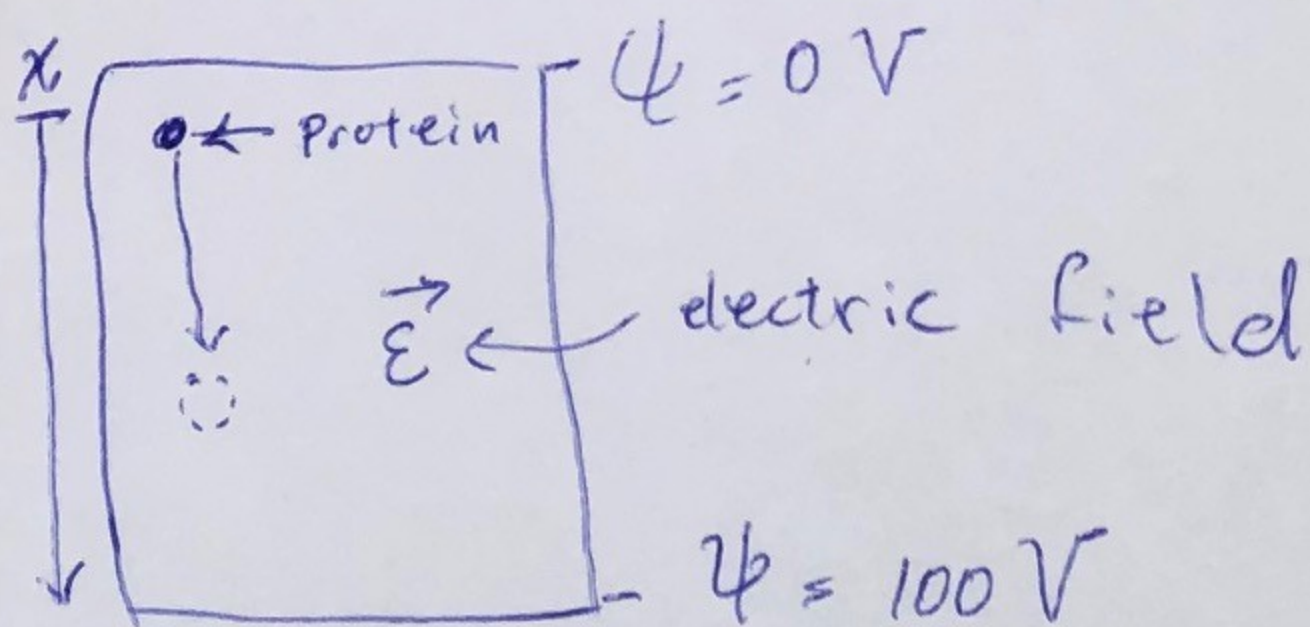
↓
3-D diffusion

$$\Rightarrow t = 6666.6 \dots \text{ seconds} \approx 1.85 \text{ hours}$$

* diffusion is also observed in gel electrophoresis

As protein or DNA sample is drifting through the gel (induced by voltage difference), the band starts to diffuse slowly. Hence, the voltage needs to be high enough that the sample does not diffuse away too much by the end of the experiment.

Drift Velocity:



From stat mech: $E = S T + \dots + \underbrace{q \psi}_{-q \cdot x \cdot E}$ ↗ voltage

for one species (protein): $G = \mu(x) = S T + \dots - q x E$

$$F = m a$$

$$F_{\text{applied}} + F_{\text{friction}} + F_{\text{random}} = m a$$

↓
electric field