

last time: more "laws" at const T

Gas: $-\frac{F}{RT} = \ln Z + E_n = \frac{h^2 n^2}{8mL^2}$ "degrees of freedom"

$\Rightarrow Z(T) = \left(\frac{2\pi m k_B T}{h^2} \right)^{3/2} \cdot V \Rightarrow E = \frac{3}{2} nRT$

$\frac{E}{3n} = \frac{1}{2} RT$ energy per degree of freedom or "equipartition"

One more example:

Diatomic molecule: how much does it "jiggle" at a given temperature T?

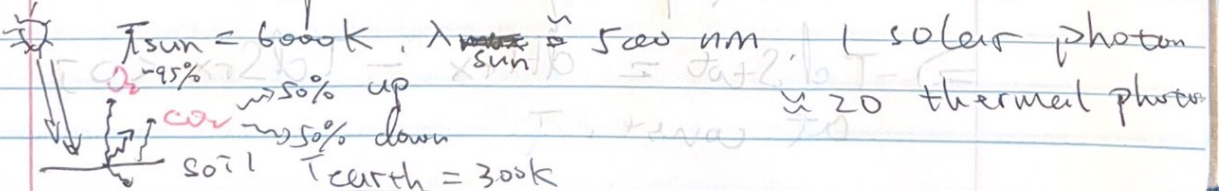
$E_n = h \cdot \nu \cdot n$ (referencing $E_0 = 0$)
 $F = k \cdot L$ force constant
 $E = TS - \int L$
 $\Rightarrow Z = \sum e^{-E_n/KT} = \sum e^{-h\nu/RT \cdot n}$

Also $F = E - TS \Rightarrow F = -fL = -kL^2 = \frac{1}{L} \frac{dE}{dL}$
 $\Rightarrow L^2 = -\frac{F}{k} = \frac{RT}{k} \ln Z$ For a isolated

Today: $Z(T)$ vs $G(T)$; Mass-action law, part I

we see $S_{tot}(t > 0) > S_{tot}(t = 0)$ or $dS_{tot} > 0$
 let's see what happen when T, P = constant

ex = HWK S. 5.1 - heat trapping in the atmosphere



HWK 55.2: photons are just like particles in a box.

$Z(T) = 8\pi \left(\frac{k_B T}{ch} \right)^3 V$ is the number of microstate of photon at temperature T in volume V per unit volume.

So far, 1 photon at 6000 K per unit volume.

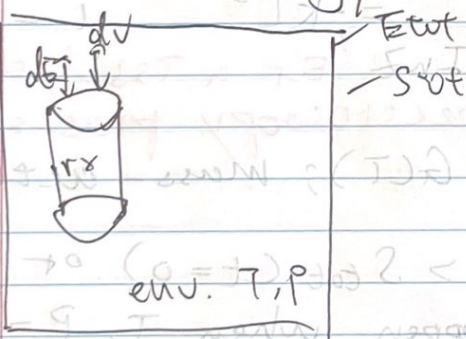
$$Z(T) = 8\pi \left(\frac{k_B \cdot 6000K}{ch} \right)^3 \approx 2 \cdot 10^{18} \text{ m}^{-3}$$

For 20 thermal photons.

$$Z(T) = \left[8\pi \left(\frac{k_B \cdot 300K}{ch} \right)^3 \right]^{20} / 20! \approx 10^{28} \text{ m}^{-3}$$

⇒ making 20 thermal photons out of 1 solar photon is very spontaneous!

Let's formulate this in terms of Free energy:



$$dS_{tot} = dS_{rx} + dS_{env}$$

$$dS_{env} = \frac{dH_{env}}{T} \quad (\text{at constant } P)$$

$$= - \frac{dH_{rx}}{T} \quad (E \text{ conservation})$$

$$\Rightarrow dS_{tot} = dS_{rx} - \frac{dH_{rx}}{T} > 0$$

multiply by $(-T)$

$$\Rightarrow -T dS_{tot} = dH_{rx} - T dS_{rx} < 0$$

at const. T

$\Rightarrow -T dS_{tot} = dG_{rx} < 0$ for a ~~spontaneous~~
spontaneous rx.