

Lecture 23 review:

1. Find one of S, E, F, G, H, \dots for your system
2. Combine with eqns. from postulates which always hold.

ex: Particles in a box

$$W = \frac{M!}{(M-N)!N!} \rightarrow S = nR \ln(V_0/N) + nR \ln V$$

$$\frac{\partial S}{\partial V} = \frac{R}{V} \rightarrow pV = nRT$$

$$\frac{\partial S}{\partial E} = \frac{1}{T} \rightarrow dS = \frac{1}{T} dE + \frac{p}{T} dV, \dots$$

ex: Heat capacity

At constant V, n : $dE = dq = C_V dt, ds = \frac{dE}{T} = \frac{C_V dt}{T}$ $C_V = T \left(\frac{\partial S}{\partial T} \right)_V$

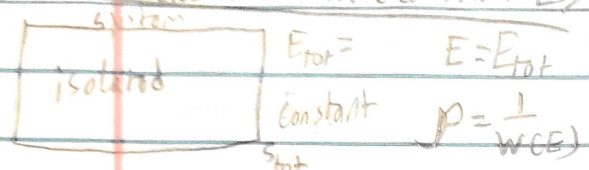
At constant P, n : $dH = dq = C_P dt, ds = \frac{dH}{T} = \frac{C_P dt}{T}$ $C_P = T \left(\frac{\partial S}{\partial T} \right)_P$

Volume concentration:

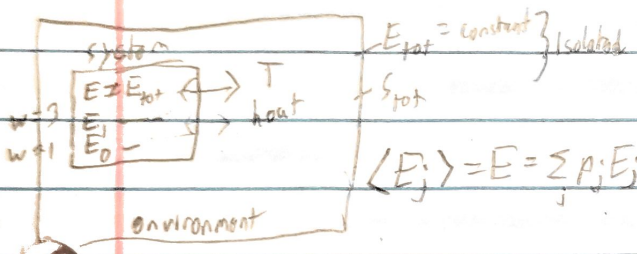
$$S = nR \left(\ln(V_0/N) + \ln V \right) = nR \left(\ln(V_0 A) + \ln V \right)$$

Lecture 24: systems at constant T, the Boltzmann factor $e^{-E_j/kT}$

Microcanonical Ensemble (constant E)



Canonical Ensemble (constant T)



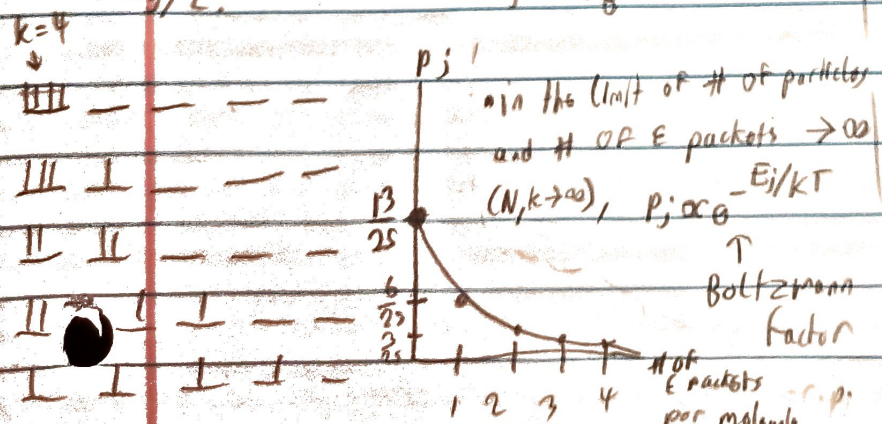
Derivation of p_j on pg. 17 of textbook

$E = \text{const.} \rightarrow p = \frac{1}{W} \quad W = e^{-TS/k_B T}$ (from $S = k_B \ln W$)

$T = \text{const.} \rightarrow p_j = \frac{W_j e^{-E_j/kT}}{Z} \quad Z = e^{-F/kT} \quad (F = E - TS)$

Illustrate with example:

- system with $N=5$ identical molecules.
- if T is the average energy per degree of freedom, $\langle E_j \rangle = k_B T$
- $K=4$ energy packets of energy ϵ .



$\langle E_j \rangle = \sum p_j E_j = \frac{13}{25} \cdot 0 + \frac{6}{25} \cdot \epsilon + \frac{3}{25} \cdot 2\epsilon + \frac{2}{25} \cdot 3\epsilon + \frac{1}{25} \cdot 4\epsilon = \frac{22\epsilon}{25}$ (1600 to normalize)

$\sum p_j = 1 \rightarrow \sum e^{-E_j/kT} = Z$

$\rightarrow p_j = \frac{e^{-E_j/kT}}{Z} \leq 1$

Notes: If a bunch of microstates have the same E , can group them.

Notes: $k_B T$ vs $R T$
 units of E (units of ϵ)
 units of ϵ per mole

$p_j = \frac{W_j e^{-E_j/kT}}{Z}$

$Z = \sum W_j e^{-E_j/kT}$
 ↓ microstates with different E

$Z = e^{-F/kT} = \sum W_j e^{-E_j/kT}$

↑ Canonical partition function counts microstates of system (like W at const. E) but difference is that at const. T all microstates (especially high E) have $p_j < 1$.