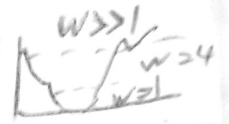


Lecture 20 review:

- The macrostate of an isolated system characterized by extensive variables E (energy), V (volume), etc... has a partition function $W(E, V, N, \dots)$ such that each microstate has equal probability $p_i = \frac{1}{W}$.

ex. particles in a box
 $W = \left(\frac{V}{v_0 N}\right)^N$ if $M = \frac{V}{v_0 N} \gg N$
 $W=1$ when $E \rightarrow 0$ (WT where)

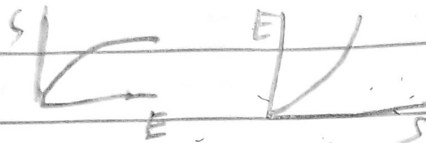


- When particles are free to move, the isolated system goes to the macrostate with largest W : $W(t>0) > W(t=0)$

We define $S = k_B \ln W$
 so $S = S_1 + S_2$, extensive

Lecture 21: Properties of S & E and intensive variables T, P, \dots

- S increases with E
 $\rightarrow E$ increases with S



in an operation $\left(\frac{\partial E}{\partial S}\right)_{V, N} > 0$
 hold constant

(like E, V, N, \dots ($E = E_1 + E_2$))

$S(t>0) > S(t=0) \rightarrow \Delta S > 0$ for an isolated system (2nd Law)

On $E \rightarrow 0$, $S = 0$ (3rd Law)

- Derivation of Energy & entropy:

$$dS = \left(\frac{1}{T_1}\right) dE_1 + \left(\frac{1}{T_2}\right) dE_2$$

ex. $y = x^2 z \Rightarrow \left(\frac{\partial y}{\partial x}\right)_z = 2xz$ or $dy = 2xz dx$
 $dS = \left(\frac{1}{T_1}\right) dE_1 - \left(\frac{1}{T_2}\right) dE_1$

$dS = 0$ when equilibrium has been reached

$$E(S, V, N, \dots) \rightarrow dE = \left(\frac{\partial E}{\partial S}\right)_{V, N} dS + \left(\frac{\partial E}{\partial V}\right)_S dV + \left(\frac{\partial E}{\partial N}\right)_{S, V} dN + \dots = \left(\frac{1}{T_1} - \frac{1}{T_2}\right) dE_1$$

define these derivatives

$$dE = T dS + P dV + \mu dN$$

$$T = \left(\frac{\partial E}{\partial S}\right)_{V, N} \Rightarrow \frac{1}{T_1} = \frac{1}{T_2} \rightarrow T_1 = T_2 \text{ at equilibrium}$$

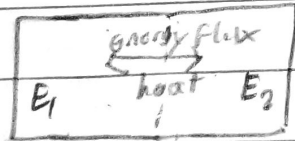
$$dS = \frac{1}{T} dE + \frac{P}{T} dV - \frac{\mu}{T} dN$$

"Temperature is the quantity

- What is T ?

isolated system, equalized when heat is allowed

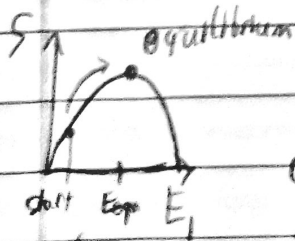
use trick of composite system



E_{tot} is constant, $dE_{tot} = 0$

to flow between systems."

analogous derivation for pressure & volume, particle # and chemical



$$dE_1 = -dE_2 \text{ (energy conservation)}$$

potential, dS

once E stops flowing (W & S have reached the biggest possible value)

$$dS = 0 = \left(\frac{\partial S}{\partial E_1}\right) dE_1 + \left(\frac{\partial S}{\partial E_2}\right) dE_2$$