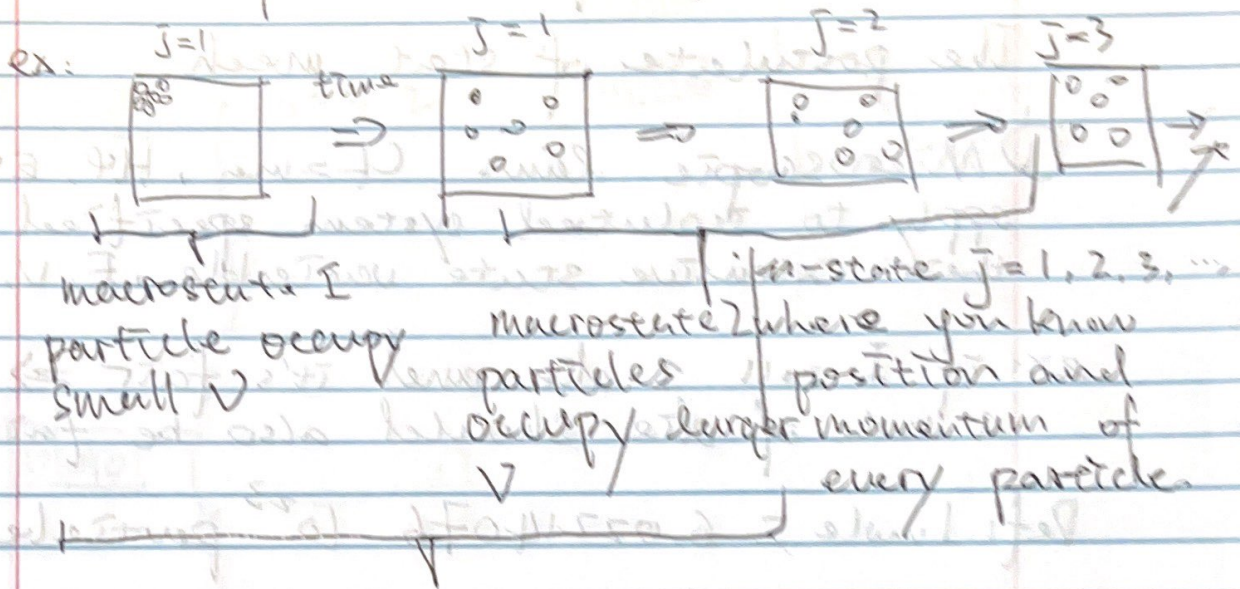


Long Time: Stat Mech - average behaviour of molecules.

States of an isolated system:



Characterized by "collective" variables called "extensive state function" eg energy =  $E$ , volume =  $V$ , number of particles  $N$ , ...

Number of accessible  $n$ -state in a microstate (size of the ensemble).

Today: The 2 postulates of stat mech  
 1<sup>st</sup> some more definition

Ensemble: Collection of all  $n$ -states consistent with a macrostate

ex: let's take dots on a rolled die = "energy"  
 $N = 2$  die,  $E = 5$

Partition function  $W$ : ensemble of states

ex:  $W = 4$ .



easy to explain.

Extensive. Scale linearly with size  $V$

Intensive: Independent with size.  $P$

The postulate of stat mech.

① Microscopic laws ( $F=ma$ ,  $H\psi$ ,  $E\psi$ ) apply to isolated system specified by their extensive state variable,  $E, V, N, \dots$

ex. if I roll 1 die and it's fair  $\Rightarrow$  rolling a mole of die should also be fair.

Def: 1 mole  $\equiv 6.02214076 \cdot 10^{23}$  particles

② Principle of equal probability:

Weak form:

(i) All microstates of a system satisfying ① have equal probability (density):

$P_j = \frac{1}{\Omega}$ . These microstate are the "microcanonical ensemble"

Strong form

(ii) for an ensemble that satisfy ②(i), "ergodic principle"

$\Rightarrow \langle P_j \rangle_{\text{ensemble}} = \langle P_j \rangle_{\text{time}}$



# Averaging and counting

ex. counting (Hwk 82.2)

$\Delta | 0 | \square$

$\square | \Delta | \square$

$\vdots$

$$6 = 3 \cdot 2$$

$\square | \Delta | \square$

$\square | \square | \Delta$

$\vdots$

$$3 = \frac{3!}{2! \cdot 1!}$$

$\square | \square | \square$

$$1 = \frac{3!}{3!}$$