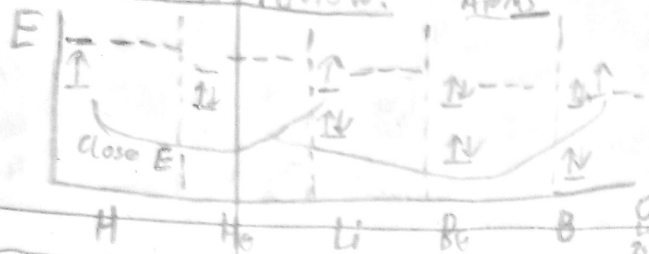


Lecture 11 review: Atoms

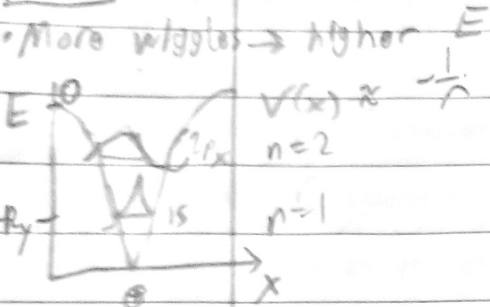


Molecules H_2^+ & O_2
 If ψ_{SA} and ψ_{SB} are wavefunctions, so are $\psi_{SA} + \psi_{SB}$ and $\psi_{SA} - \psi_{SB}$
 Far Far close

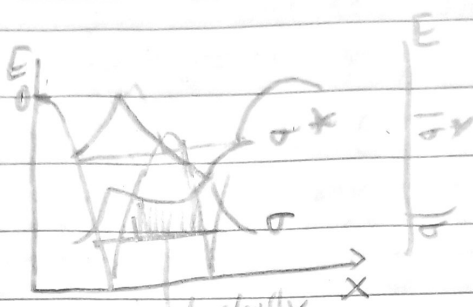
Lecture 12: Bonding, Forbidden Region & quantum interference

of the 4 quantum #s, at least 1 must be different.

Atom



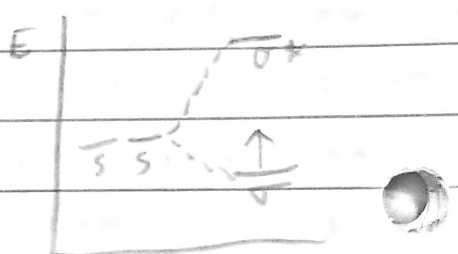
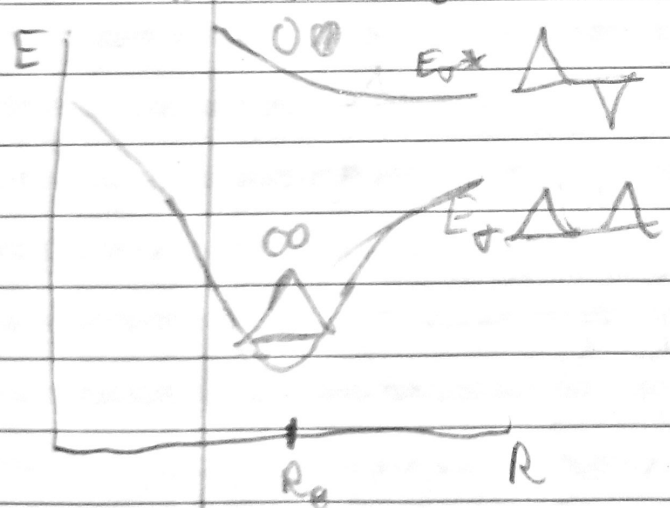
Molecule



$\psi + \psi$ adds
 $\psi - \psi$ subtracts
 quantum interference "cancellation"

• bonding is additive quantum interference

How does the energy of orbitals depend on nuclear distance R ?



If e^- were in σ^* ,
 (higher E wavefunction/orbital),
 the molecule dissociates.